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Research Article

SPECTROPHOTOMETRIC METHODS FOR SIMULTANEOUS ESTIMATION OF METFORMIN HCL AND FENOFIBRATE IN THEIR SYNTHETIC MIXTURE

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ABSTRACT

Two simple spectrophotometric methods have been developed for simultaneous estimation of Metformin HCl and Fenofibrate from their synthetic mixture. Method-I involved simultaneous equation method and Method-II is the Q-absorbance method. For simultaneous equation method, the absorbances of the standard solutions were taken at two wavelengths 237 nm (λ -max of Metformin HCl) and 288 nm (λ -max of Fenofibrate). For Q-absorbance method, the absorbances of the standard solutions were taken at two wavelengths 237 nm (λ -max of Metformin HCl) and 249 nm (Isoabsorptive point), in methanol. Linearity range was found to be 3-20 µg/ml for Metformin HCl and Fenofibrate in both methods based on the ratio of the two drugs in combined dosage form. The accuracy and precision of the methods were determined and validated statistically. Both methods showed good reproducibility and recovery with RSD less than 2. Proposed methods were found to be rapid, specific, precise and accurate and can be successfully applied for the routine analysis of Metformin HCl and Fenofibrate in pharmaceutical dosage form.

Keywords: Metformin HCl, Fenofibrate, Simultaneous equation method, Q-absorbance method

INTRODUCTION

Metformin HCl (MET) chemically, Hydrochloride salt of N,N-dimethylimidodicarbonimidic diamide¹. It is antidiabetic drug, primarily acts by decreasing intestinal absorption of glucose reducing hepatic glucose production and increasing sensitivity thus enhancing peripheral uptake utilization of glucose². Fenofibrate (FENO) chemically, 2-[4-(4-chlorobenzoyl)phenoxy]-2-methyl-propanoic 1acid, methylethyl ester³. It is lipid lowering agent which lowers plasma triglyceride be enhancing lipoprotein lipase synthesis thus increasing very low density lipoprotein (VLDL) catabolish with consequent increase in high density lipoprotein⁴. The fixed dose combination increased patient convenience and improved compliance for patients already stabilized on two medications, FENO and MET which can control both the blood lipids and blood glucose of patients.

MET is official in IP⁵, BP⁶, and USP⁷ while FENO is official in BP⁵. A deep Literature survey shows that combination of these two drugs is not official in any pharmacopoeia and no official or reported method is available for simultaneous estimation of MET and FENO in synthetic mixture. Various reported methods are available for estimation of MET alone such as UV⁸, HPLC⁹⁻¹³, LC/MS¹⁴⁻¹⁶, HPTLC¹⁷, Capillary Electrophoresis¹⁸, Voltametry¹⁹ and in combination with other drugs like UV²⁰⁻²⁴, HPLC²⁴⁻³⁵, HPTLC³⁶, LC/MS³⁷⁻³⁹, Capillary Zone Electrophoresis⁴⁰ while for estimation of FENO alone such as UV⁴¹, HPLC⁴²⁻⁴⁶, UPLC⁴⁷, LC-MS/MS^{48,49} and in combination with other drugs like HPTLC⁵⁰. In the present investigation an attempt has been made to develop simple, rapid, economic and accurate spectrophotometric method for simultaneous estimation of MET and FENO from their synthetic mixture.

MATERIALS AND METHODS

Instrumentation and Apparatus

A shimadzu model 1600 (Japan) double beam UV/Visible spectrophotometer with spectral width of 2 nm, wavelength accuracy of 0.5 nm and a pair of 10 mm matched quartz cell was used to measure absorbance of all the solutions. Spectra

were automatically obtained by UV-Probe system software (UV Probe version 2.10). A Sartorius CP224S analytical balance (Gottingen, Germany), an ultrasonic bath (Frontline FS 4, Mumbai, India) was used in the study.

Reagents and Chemicals

Kindly gifted reference standards of MET (Torrent Pharmaceutical Ltd, Gujarat, India) and FENO (INTAS Pharmaceutical Ltd, Gujarat, India), Laboratory prepared synthetic mixture of MET and FENO, Methanol (AR Grade, S. D. Fine Chemicals Ltd., Mumbai, India) and Whatman filter paper no. 41 (Millipore, USA) were used in the study.

Preparation of solutions and mixture

Preparation of standard stock solutions:

accurately weighed (10 mg) of a standard drugs were transferred in two different 100 mL volumetric flasks, dissolved and diluted in methanol to get 100 μ g/mL standard stock solutions each of MET and FENO.

Preparation of Synthetic mixture of FENO & MET:

Placebo powder mixture was prepared by mixing 2 gm magnesium stearate, 2 gm starch, 10 gm talc and 6 gm lactose. Synthetic mixture was prepared by mixing 1 gm placebo powder mixture with 160 mg FENO and 500 mg MET.

Wavelength selection

The working standard solutions of MET and FENO, each of 10 µg/mL were scanned separately in the UV range of 200-400 nm. Spectrum data were recorded at an interval of 1 nm. From the absorption overlain spectra of the both drugs, different wavelengths i.e. $\lambda_1 = 237 \text{ nm} (\lambda_{max} \text{ of MET})$ and $\lambda_2 = 288 \text{ nm} (\lambda_{max} \text{ of FENO})$ for Method-I and $\lambda_1 = 237 \text{ nm} (\lambda_{max} \text{ of Met})$ and $\lambda_3 = 249 \text{ nm}$ (Isoabsorptive point) for Method-II were selected.

Preparation of calibration curve

From the standard stock solutions, 0.3, 0.5, 1.0, 1.5 and 2.0 mL aliquots of both drugs were transferred in two different series of 10 mL volumetric flasks and volumes were made up to mark with methanol to get working standards having concentration in the range of 3–20 μ g/mL for both drugs.

Absorbances of each working standard solution of both drugs were measured at λ_1 , λ_2 and λ_3 .

In Method-I, three Calibration curves of absorbance Vs concentration were constructed, of which one for MET at $\lambda_{1,}$ while two for FENO at λ_{1} and λ_{2} . At λ_{2} , MET have zero absorbance for any concentration.

In Method-II, three Calibration curves of absorbance Vs concentration were constructed of which two at λ_1 for MET and FENO, while one at λ_3 (i.e. Isoabsorptive point of both drug). From respective calibration curve of both drugs absorptivity values were calculated and used in further calculation of the concentration of both drugs in the synthetic mixture.

Table 1: Regression A	nalysis Data and Summar	y of Validation Paramete	r for the pro	posed Method-I & II.

		Method-I		Method-II				
Parameters	MET at FENO at		FENO at	MET at	FENO at	MET & FENO		
	237 nm	237 nm	288 nm	237 nm	237 nm	at 249 nm		
Concentration range (µg/mL)	3 - 20	3 - 20	3 - 20	3 - 20	3 - 20	3 - 20		
Molar absorptivity (1 mole ⁻¹ cm ⁻¹)	2.984	3.254	3.186	3.129	2.864	2.769		
Regression equation (y=mx+c)								
Slope (m)	0.0985	0.0237	0.0508	0.0982	0.0240	0.0322		
Intercept (c)	0.0063	0.0239	0.0084	0.0060	0.0269	0.0108		
Correlation coefficient (r ²)	0.9973	0.9977	0.9998	0.9978	0.9989	0.9988		
Robustness (CV, %)	0.58	0.79	0.34	0.63	0.86	1.08		
LOD (µg/mL)	0.5	0.5	0.5	0.5	0.5	0.5		
LOQ (µg/mL)	1	1	1	1	1	1		
Repeatability (n=5) (CV, %)	0.59	1.43	0.52	0.42	0.98	1.04		
Precision (CV, %)								
Interday (n =5) (CV, %)	0.47-1.93	0.38-2.06	0.58 -1.63	0.52 -1.85	0.35-1.83	0.58- 1.90		
Intraday (n =5) (CV, %)	0.56- 1.75	0.32-1.48	0.38- 1.58	0.35-1.78	0.28- 1.45	0.32-1.65		

n = number of determinations

Table 2: Results of recovery study by proposed methods

			Std. reco	vered (mg)	Mean % Recovery ± S.D (n=3)		
Drug	Level	Standard added	Method I (mg) Method II (mg)		Method I	Method II	
		Method I & II (mg)					
	Ι	500	495	506	99 ± 1.55	101.20±1.25	
MET	II	1000	987	1026	98.7 ± 1.86	102.6±1.38	
	III	1500	1492	1487	99.40 ± 1.32	99.13±1.55	
	Ι	80	78	82	97.50 ± 1.45	102.5±1.12	
FENO	II	160	156	162	97.30 ± 1.18	101.2±1.24	
	III	240	244	236	101.6 ± 0.85	98.3±0.98	

n = number of determinations

Table 3: Intra-Day Precision data for analysis of MET and FENO by Proposed method (n=5)

Conc.	MET at 237 nm		FENO at 237 nm		FENO at 288 nm		MET & FENO at 249 nm	
(µg/mL)	Mean ±	% CV	Mean ±	% CV	Mean ±	% CV	Mean ±	% CV
	S.D. (n=5)		S.D. (n=5)		S.D. (n=5)		S.D. (n=5)	
3	$0.272 \pm$	1.38	$0.022 \pm$	1.18	0.104 ±	0.38	$0.055 \pm$	1.32
	0.005		0.003		0.002		0.004	
5	$0.468 \pm$	0.56	$0.074 \pm$	0.88	0.192 ±	0.68	0.132 ±	0.95
	0.006		0.004		0.006		0.002	
10	0.915 ±	1.65	0.148 ±	1.48	0.386 ±	1.15	0.239 ±	0.32
	0.004		0.002		0.003		0.004	
15	$1.412 \pm$	1.75	$0.376 \pm$	0.32	0.590	1.49	$0.379 \pm$	1.28
	0.004		0.005		±0.005		0.003	
20	$1.752 \pm$	0.83	0.512 ±	0.64	0.808	1.58	0.511 ±	1.65
	0.002		0.003		±0.004		0.004	

n = number of determinations

Table 4: Inter-Day Precision data for analysis of MET and FENO by Proposed method (n=5)

Conc.	MET at 237 nm		FENO at 237 nm		FENO at 288 nm		MET & FENO at 249 nm	
(µg/mL)	Mean ±	% CV	Mean ± S.D.	% CV	Mean ± S.D.	% CV	Mean ± S.D.	% CV
	S.D. (n=5)		(n=5)		(n=5)		(n=5)	
3	$0.267 \pm$	1.58	$0.020 \pm$	1.15	0.101 ±	0.58	0.052 ± 0.006	1.90
	0.003		0.004		0.005			
5	$0.458 \pm$	0.47	$0.072 \pm$	0.38	0.197 ±	0.77	0.128 ± 0.003	0.85
	0.005		0.002		0.005			
10	0.911 ±	1.65	0.146 ±	1.47	0.381 ±	1.45	0.237 ± 0.005	0.58
	0.006		0.004		0.002			
15	1.409 ±	1.93	0.376 ±	2.06	0.594 ±	1.29	0.376 ± 0.004	1.58
	0.005		0.005		0.006			
20	1.754 ±	0.94	0.517 ±	0.74	$0.805 \pm$	1.63	0.517 ± 0.006	1.38
	0.003		0.005		0.003			

n = number of determinations



Figure 1: Overlain spectra of MET and FENO

RESULTS AND DISCUSSION Method Development⁵¹

From the absorption overlain spectra of standard MET and FENO, the selected wavelengths are shown in Figure1. The overlay spectra pattern of both drug suggests that at $\lambda_{2,MET}$ have zero absorbance for any concentration there is sufficient distance between λ_{max} of both drugs, the criteria run obtaining maximum precision by simultaneous equation method were calculated and found to be outside the range 0.1-2.0. The Isoabsorptive point was found to be 249 nm which was used in Q-Absorbance ratio method.

Calibration curves for MET at λ_1 while for FENO at λ_1 and λ_2 for Method-I, for Method-II at λ_1 and λ_3 for both drugs were constructed and Beer's law range was found to be 3-20 µg/ml at all selected wavelengths for both Methods.

From the respective calibration curves and regression equation, the calculated absorptivity value were found to be $ax_1 = 950$, $ax_2 = 0$, $ay_1 = 240$, $ay_2 = 508$ for Method-I and $a_1x = 890$, $a_2x = a_2y = 270$, $a_1y = 185$ for Method –II.

The calculated absorptivity values at particular wavelength were substituted in the equations for Method-I and II and concentration of both drugs from synthetic mixture was found out.

For Method-I:

 $\begin{array}{lll} Cx = & (A_1 ay_2 - A_2 ay_1) / (ax_1 ay_2 - ax_2 ay_1) & (1) \\ Cy = & (ax_1 A_2 - ax_2 A_1) / (ax_1 ay_2 - ax_2 ay_1) & 2) \\ \text{Where,} \\ C_X & \text{and} & C_Y & = & \text{the concentration of MET and FENO,} \\ \text{respectively.} \\ A_1 = & \text{absorbance of mixture at } \lambda_1 \\ A_2 = & \text{absorbance of mixture at } \lambda_2 \\ ax_1 = & \text{absorptivity value of the MET at } \lambda_1 \\ ax_2 = & \text{absorptivity value of the MET at } \lambda_2 \\ ay_{1=} & \text{absorptivity value of the FENO at } \lambda_1 \\ ay_{2=} & \text{absorptivity value of the FENO at } \lambda_2 \\ \end{array}$

For Method-II:

 $\begin{array}{ll} C_{\rm X} = (Q_{\rm m} - Q_{\rm y}) \times A/(Q_{\rm X} - Q_{\rm Y}) \times ax_1 &(3) \\ C_{\rm Y} = (A/ax_1) - C_{\rm X} & (4) \end{array}$

Where,

 C_X and C_Y = the concentration of MET and FENO, respectively.

A = the absorbance of mixture at Isoabsorptive point (λ_3)

 $ax_1 = absorptivity of MET at Isoabsorptive point (\lambda_3)$

 Q_X = (absorptivity at Isoabsorptive point, λ_3)/(absorptivity of MET at λ_1)

 $Q_{\rm Y}$ = (absorptivity at Isoabsorptive point, λ_3)/(absorptivity of FENOe at λ_2)

 Q_m = (absorbance of the FENO at $\lambda_2)/(absorbance of the MET at <math display="inline">\lambda_1)$

Validation of Method⁵²

Linearity:

Both methods were found to be linear over the range of 3-20 μ g/mL for the both drugs at all selected wavelengths with the values of correlation coefficient (r²) > 99. The regression analysis data and optical parameters of both methods were shown in table 1.

Accuracy:

To check the accuracy of proposed methods, multilevel recovery study by standard addition method in placebo powder mixture was carried out. After mixing spiked standard drug at each level were ultrasonicated for 30 min and extracted in methanol. The solutions were filtered through Whatman filter paper No. 41. Suitable aliquots of filter at each level were diluted to get final solutions at each level. Absorbances of final solutions were measured at selected wavelengths for Method I & II and concentrations of both drugs were calculated by using respective regression equations. The results of multilevel recovery study for both methods were shown in Table 2. From the results it can be concluded that both developed methods were accurate.

Method Precision (Repeatability):

Repeatability of measurement of absorbance at selected wavelengths for both drugs using both method were evaluated using 5 replicates of the same concentration (10 μ g/mL of MET and FENO). The calculated CV, % value for method precise was found to be within limit as shown in Table 1, suggested that both methods were repeatable.

Intermediate Precision:

The intra-day and inter-day variation of both methods were evaluated at 5 different concentration levels (3, 5, 10, 15 and 20 μ g/mL) for both drugs. The percentage Co-efficient of Variance (CV, %) values of within-day and day-to-day were calculated. The results of intra and inter-day variation are depicted in Table 3 and 4, respectively. The CV, % value for day to day and within day was found to be within limit suggested that both methods were sufficiently precise over the calibration range.

Limit of detection (LOD) and Limit of quantitation (LOQ):

LOD was checked by visual method as per ICH guidelines. The values of LOD and LOQ were found to be 0.5 and 1.0 μ g/mL (Table 1), for both drugs in both developed methods reveals that purposed method can be applied at low concentration level with sufficient sensitivity.

Robustness:

The study was carried out by change in selected wavelengths for 0 ± 0.1 , 0.2, 0.3, 0.4 and 0.5 nm and the CV, % values were calculated, which were found to be less than 2 %. The results suggested that method was robust under experimental conditions.

CONCLUSION

Both developed methods have linear response in the stated range for both drugs and are accurate, precise simple and rapid. The developed methods can be readily carried out at laboratory level and small scale industries using inexpensive instrument i.e. UV/Visible Spectrophotometer. In Future, both methods can be routinely used as quality control of MET and FENO from their combined dosage forms.

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